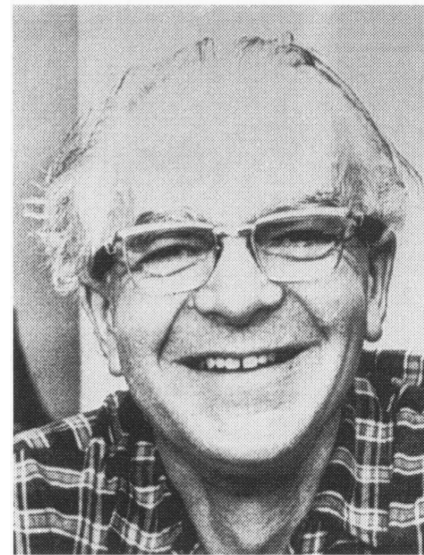


BioChem 330 - Course Outline

- **Metabolism and Bioenergetics (II)**
 - ENZYME CATALYSIS:
 - kinetic constants k_{cat} , K_m
 - Catalytic strategies, the serine proteases
 - CATABOLISM (*breakdown*)
 - Carbohydrates
 - Glycolysis
 - Tricarboxylic Acid Cycle
 - Electron Transport
 - Chemiosmosis and ATPase
 - Fatty acids and amino acids

Overview for Today, November 17, 2011

- Coupling proton pumping to ATP production through the electrochemical potential
- Smallest little motor in the world, the F_0F_1 ATPase



Peter Mitchell
1920–1992

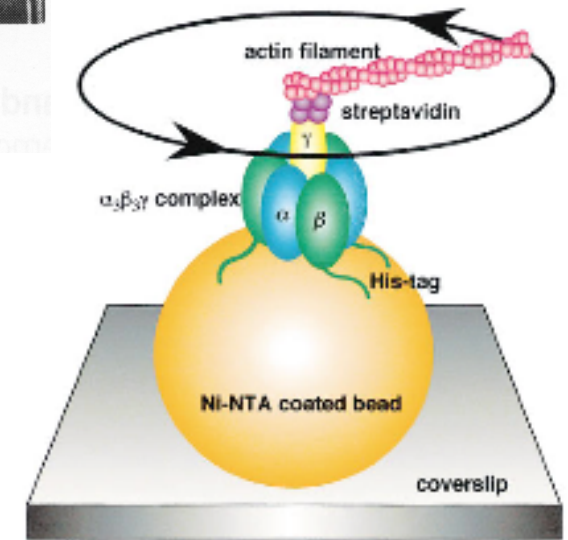


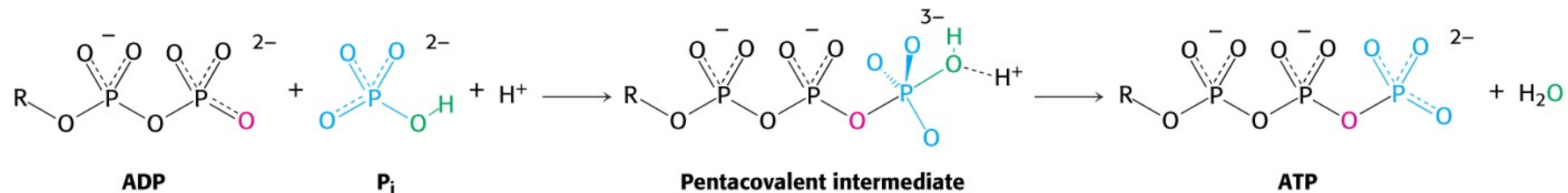
Figure 1. Experimental System (Not to Scale)

ET coupled to ATP synthesis

- ET chain induces electrochemical potential gradient by pumping protons across energy transducing inner mitochondrial membrane against proton and voltage gradient:

$$\begin{aligned}
 & - \qquad \qquad \qquad \Delta pH = 1.4 \qquad \qquad \qquad \Delta V = 0.14 \text{ V} \\
 & - \Delta G = 2.303 RT \Delta pH \qquad \qquad \qquad + Z F \Delta V =
 \end{aligned}$$

- ATP synthesis takes those protons pumped by ET and allows them to flow back into the cell through a clever little motor known as ATP synthase, catalyzes:

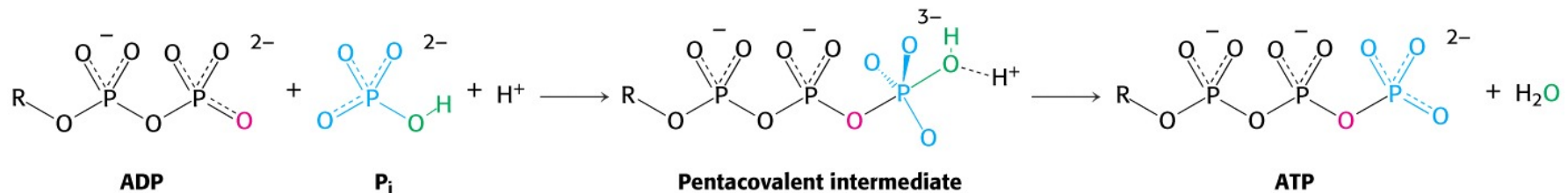


ET coupled to ATP synthesis

- ET chain induces electrochemical potential gradient by pumping protons across energy transducing inner mitochondrial membrane against proton and voltage gradient:

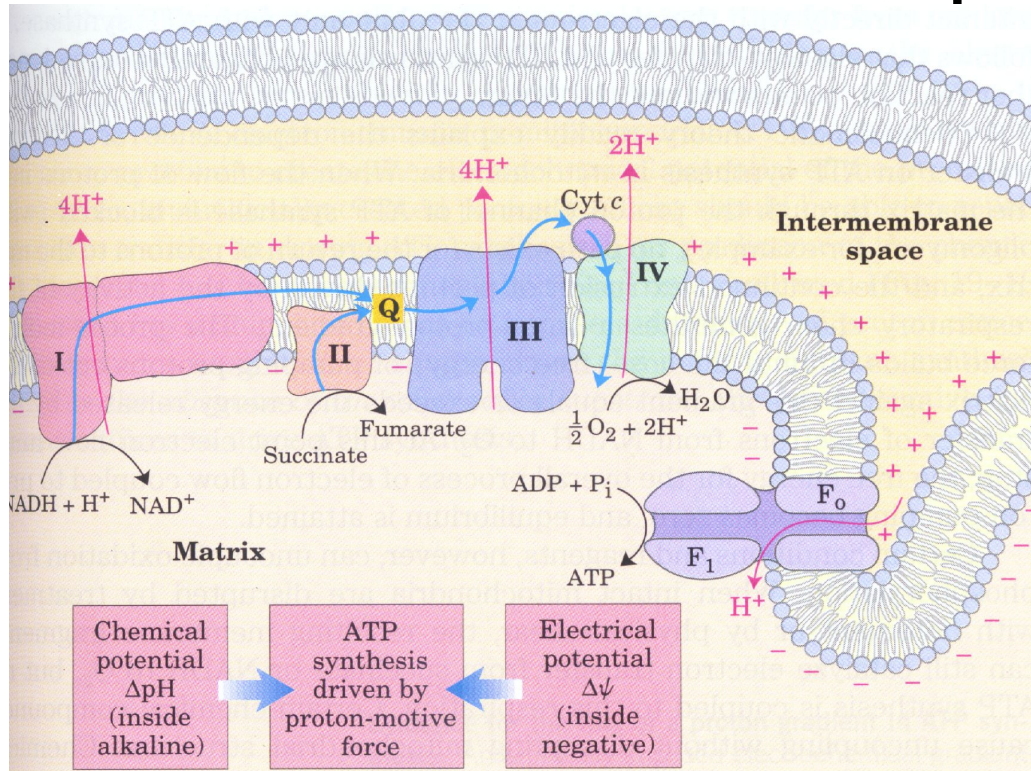
$$\begin{aligned}
 & - \quad \Delta pH = 1.4 \quad \Delta V = 0.14 \text{ V} \\
 & - \quad \Delta G = 2.303 RT \Delta pH + Z F \Delta V = \boxed{5.2 \text{ kcal/mole H}}
 \end{aligned}$$

- ATP synthesis takes those protons pumped by ET and allows them to flow back into the cell through a clever little motor known as ATP synthase, catalyzes:



$$\Delta G^{\circ'} = + 7.3 \text{ kcal/mole}$$

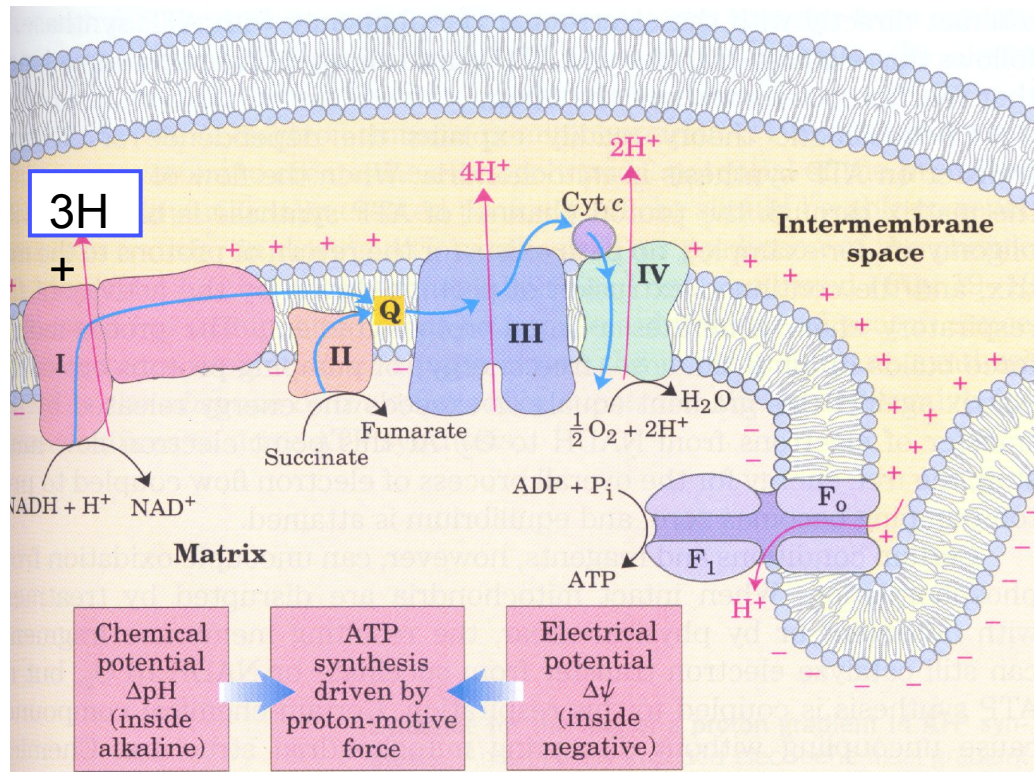
Chemiosmosis: the picture



- Understanding the detailed book-keeping of H^+ stoichiometry has changed over time, your text's accounting at odds with newer texts which cite 3H^+ through Complex I

• The intermembrane space is essentially in equilibrium with the cytoplasm, which is buffered at pH higher than the mito, and so, added protons don't change the pH of the cytoplasm but have profound effect on the pH of the matrix (like adding a few drops of acid to the ocean would have no effect, but adding it to a thimble would have a large effect).

Chemiosmosis: the picture



- Electron flow is accompanied by proton transfer across the mitochondrial membrane, producing both a chemical (ΔpH) and an electrical ($\Delta\psi$) gradient

- The proton-motive force that drives protons back into the matrix provides the energy for ATP synthesis, catalyzed by the F1 complex associated with F₀.

- $\Delta G = 2.3RT \Delta\text{pH} + F \Delta\psi$

- $F = 23.06 \text{ kcal/mole/V}$

- **Mitochondria:**

- $\Delta\text{pH} = 1$ (alkaline inside)

- $\Delta\psi = 0.1\text{-}0.2 \text{ V}$ (inside negative)

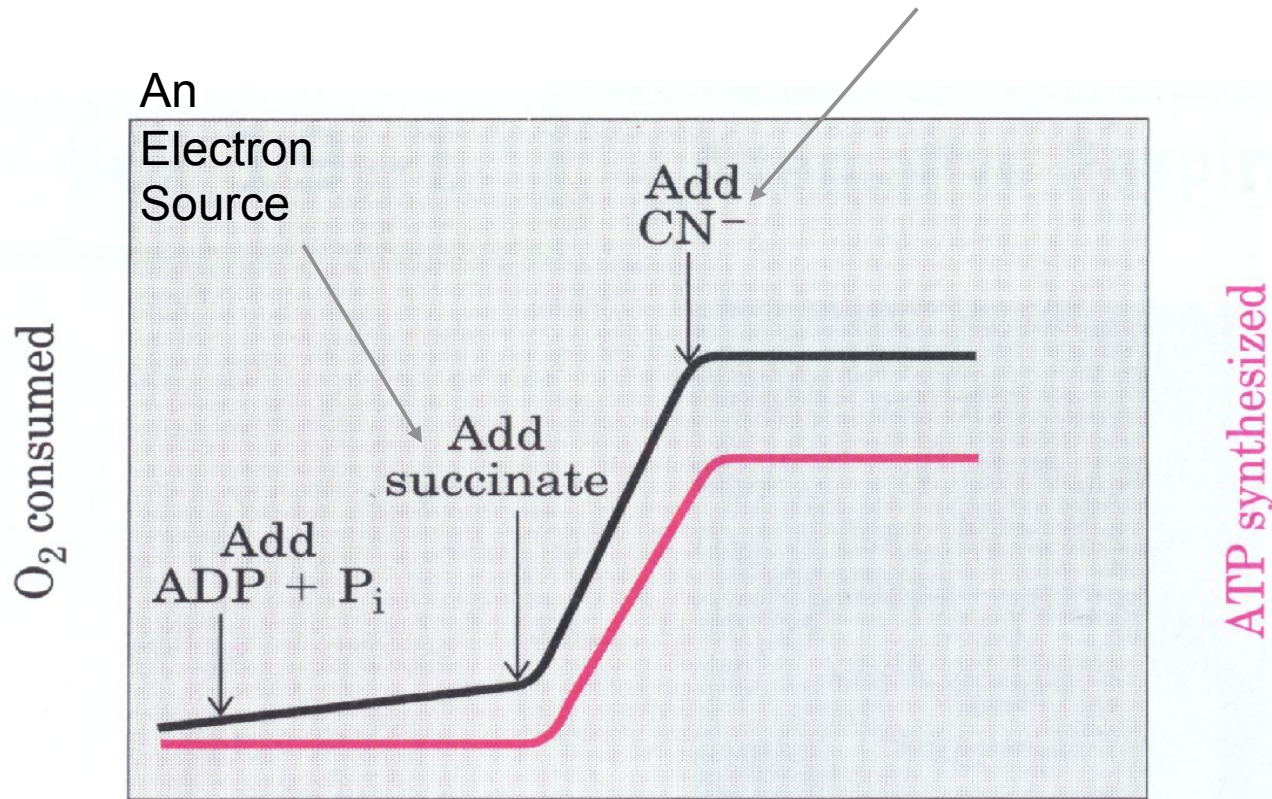
Basic evidence

- Electron transport is *coupled* to ATP synthesis (more detail in a moment)
- Coupling requires an intact membrane
- Electron flow through individual complexes results in proton pumping
- Proton carriers *uncouple* electron transport and ATP synthesis

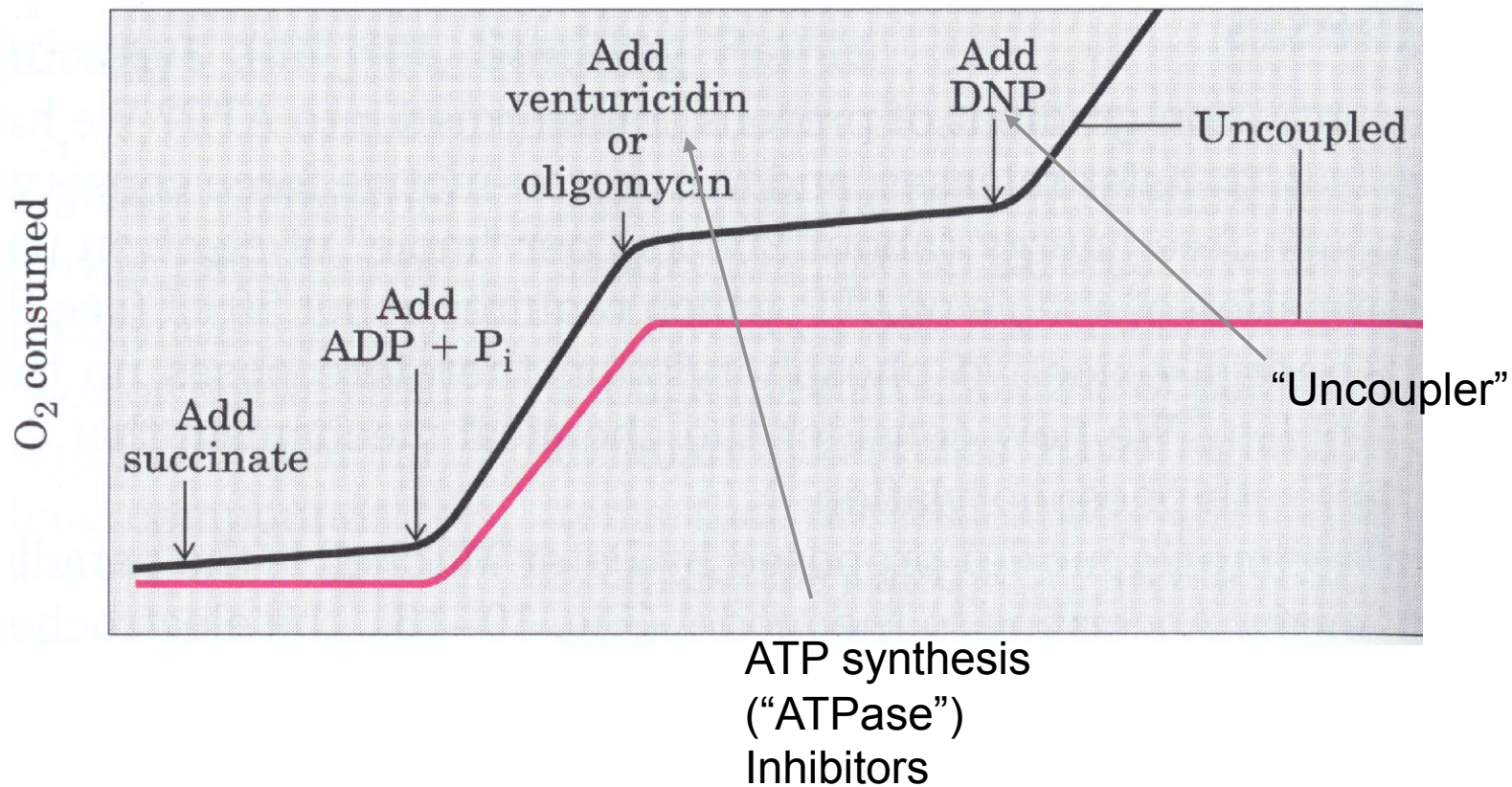
What experiments look like...

Coupling:

An inhibitor
Of electron transport
(Binds to cytochrome oxidase)

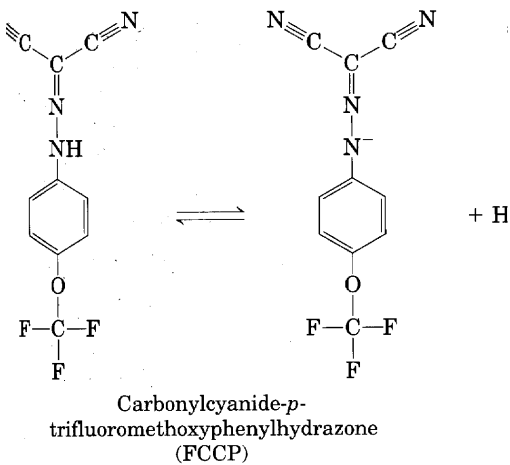
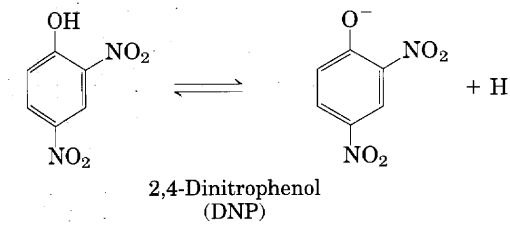


Or, more subtly...

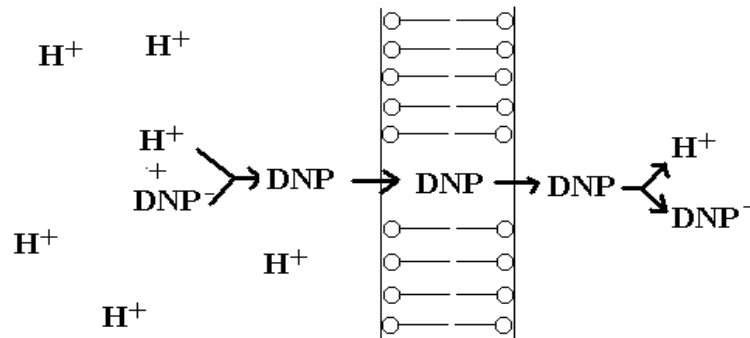


“Uncouplers”

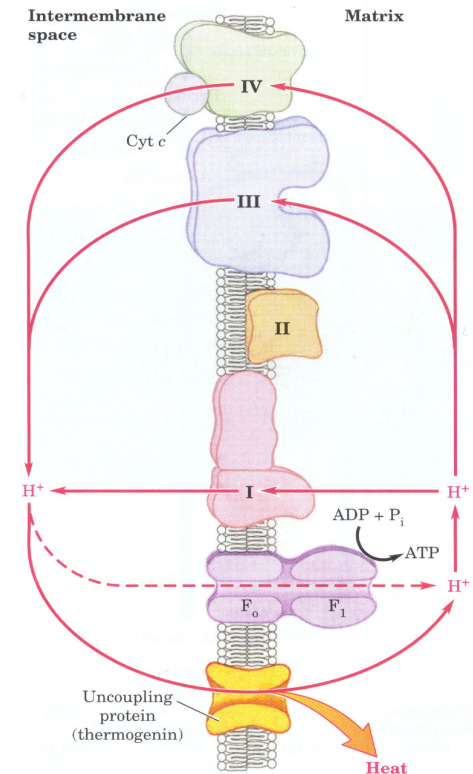
What they look like...



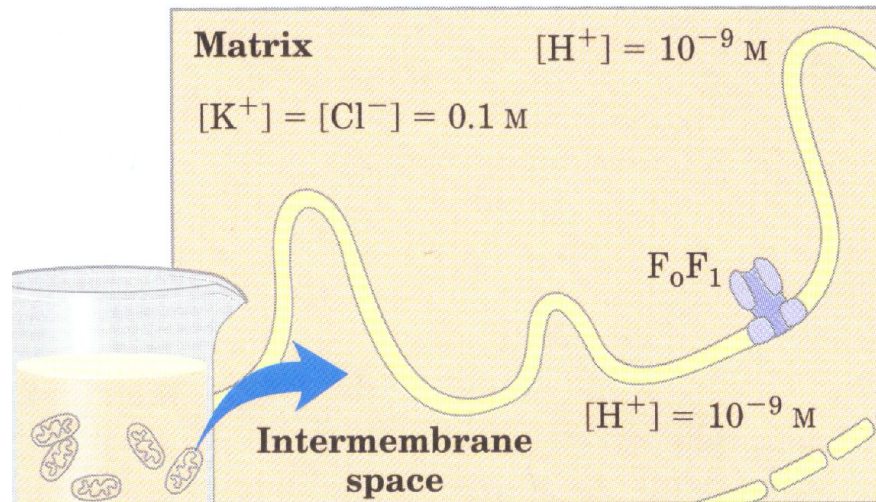
What they do...



A natural version of these things in brown fat...



Creation of an electrochemical gradient is sufficient...



(a)

pH lowered from 9 to 7;
valinomycin present; no K^+

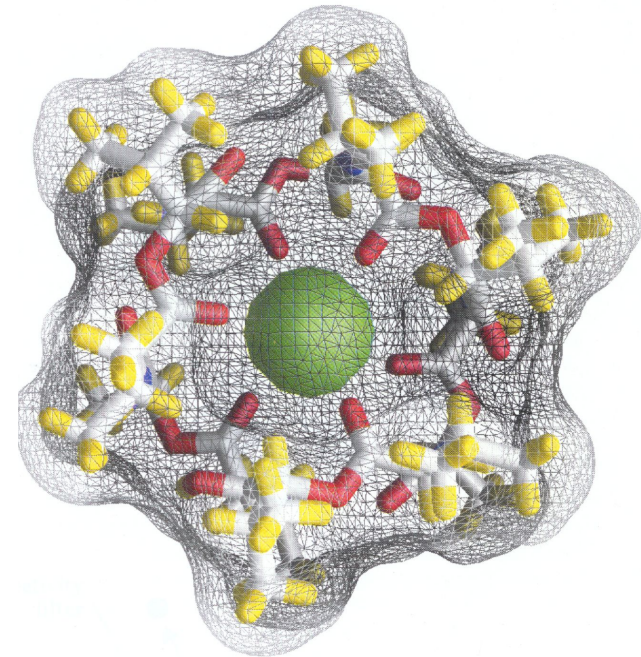
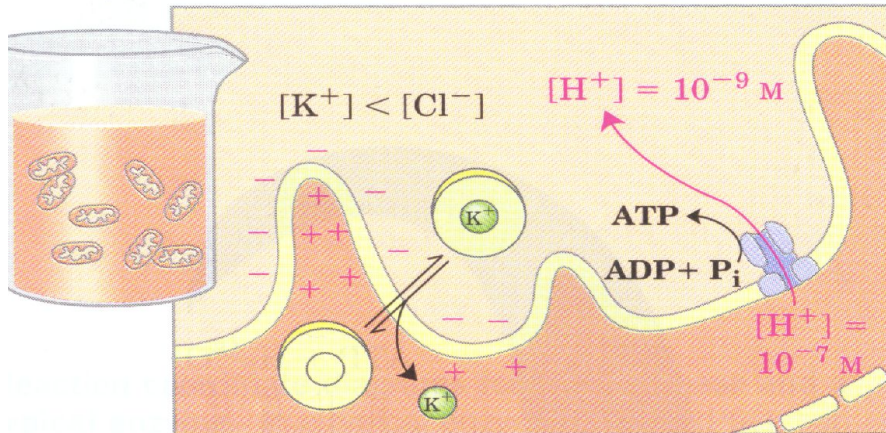
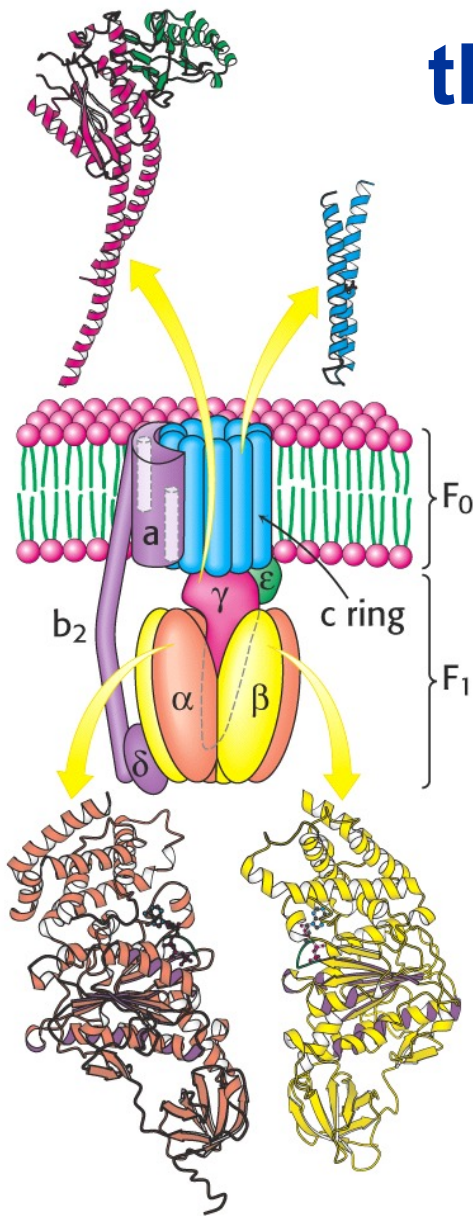


figure 12-37

Valinomycin, a peptide ionophore that binds K^+ . In this image, the surface contours are shown as a transparent

Structure of F_0F_1 ATPase: the littlest motor in the world



1. F_0 unit: proton conducting unit, integral membrane proteins, proton channel formed by a ring of 10-14 c subunits and a single a subunit which binds to outside of c ring.
2. F_1 catalytic unit ($\alpha_3\beta_3\gamma\delta\epsilon$) which consists of a hexamer of alternating α and β subunits (only β is active), a central stalk of γ and ϵ long α -helical proteins, which secure F_1 to F_0 . Assymetry of γ is key to activity
3. b subunit of F_0 interacts with F_1 through d .

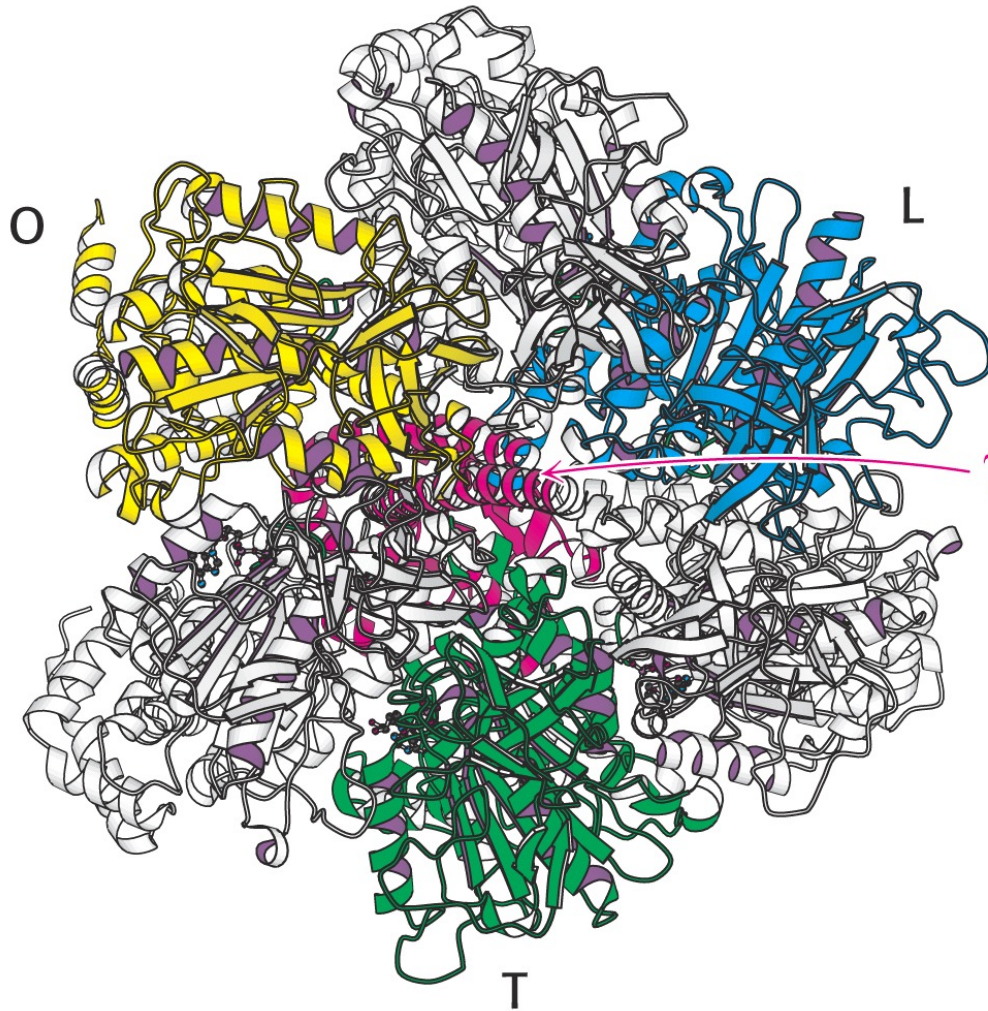
Looking down through the top of F1

* γ subunit breaks the symmetry, interacting differently with each of the β subunits (color),

• ATP binds differently to each conformer:

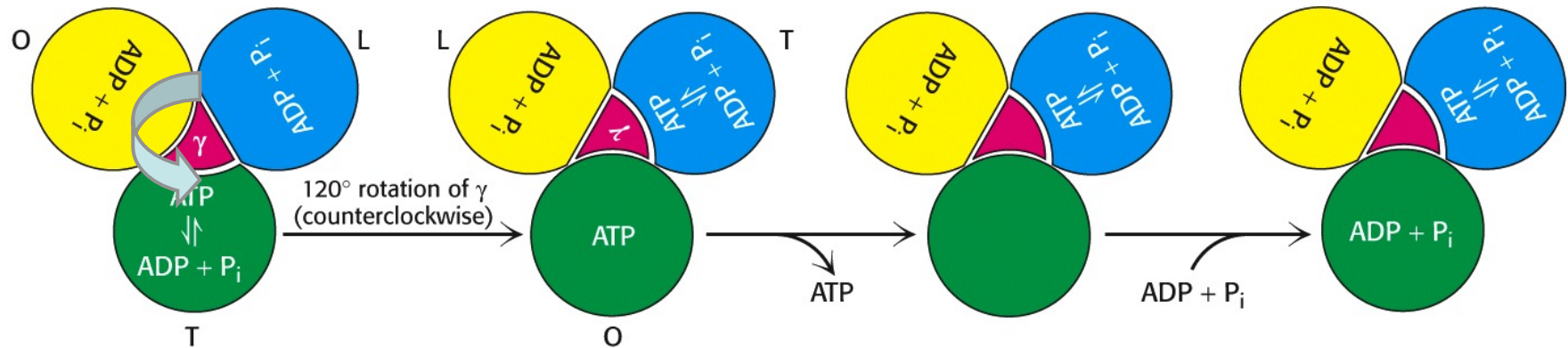
• γ subunit

- L is loose (ADP and Pi bound),
- T is tight (ATP made),
- O is open, (bound ATP released).



• Intact assembly, no proton gradient, ATP is formed, but not released.

Binding Change Model for coupling



Perspective from matrix above F1 : rotation of γ counterclockwise ATP synthesis

1. Each **counterclockwise step** of the red wedge (γ subunit) by 120° changes β conformation: $O \rightarrow L$ or $L \rightarrow T$, or $T \rightarrow O$
2. Each full cycle of γ ($3 \times 120^\circ$ steps) makes 3ATP (1 from each β)
3. Catalysis only happens in T
4. Motor can go backwards for hydrolysis (ATPase) in bacteria, but only ATP synthesis in mitochondria and chloroplasts.

F1 the first single molecule experiments

Reporter gene assays. THP-1 cells were transfected with DEAE-Dextran[™] and RAW 264.7 cells with calcium phosphate²⁷. Cells were transfected with a β -galactosidase expression vector pCMV- β , with the reporter plasmid pG5E1-bLuc, with an expression vector encoding a GAL4-MEF2C fusion protein or GAL4(1-147) or GAL4-MEF2C mutants, and with the expression vector encoding a constitutively active form of MKK6b(E) or with empty vector pcDNA3. In some experiments the cells were also transfected with increasing amounts of DNA (0, 2, 4, 8, 12 or 16 μ g) from an expression plasmid for p38(M). The total amount of DNA for each transfection was kept constant using pcDNA3. In studies using FHIP1, the inhibitor was added (3 μ M) 36 h after transfection for 1 h; and the cells were then treated with or without LPS for 8 h. LPS (5 μ g ml⁻¹ and 10 ng ml⁻¹) was used to stimulate THP-1 and RAW 264.7 cells, respectively. The relative luciferase activities presented were normalized by dividing the luciferase activity by β -galactosidase activity.

Phosphoamino-acid analysis and phosphopeptide mapping. These methods were performed as described¹⁴. GAL4-MEF2C fusion protein and mutants from [³²P]orthophosphate-labelled permanently transfected RAW 264.7 cells (1 mCi ml⁻¹, 2 h) treated with or without LPS (10 ng ml⁻¹, 2 h) were immunoprecipitated with anti-GAL4 DNA-binding domain monoclonal antibody RKSC1 (Santa Cruz).

Received 18 November 1996; accepted 21 January 1997.

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Direct observation of the rotation of F₁-ATPase

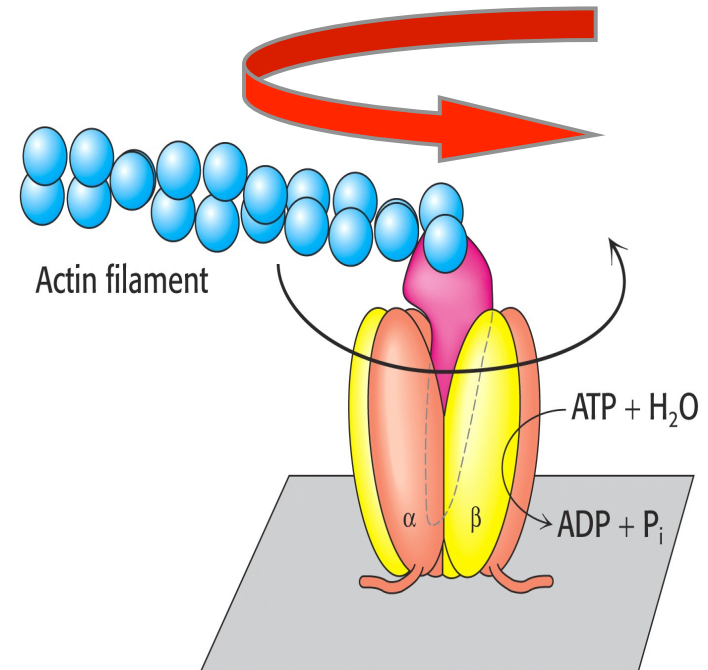
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Cells employ a variety of linear motors, such as myosin¹⁻³, kinesin⁴ and RNA polymerase⁵, which move along and exert force on a filamentous structure. But only one rotary motor has been investigated in detail, the bacterial flagellum⁶ (a complex of about 100 protein molecules⁷). We now show that a single molecule of F₁-ATPase acts as a rotary motor, the smallest known, by direct observation of its motion. A central rotor of radius ~1 nm, formed by its γ -subunit, turns in a stator barrel of radius ~5 nm formed by three α - and three β -subunits⁸. F₁-ATPase, together with the membrane-embedded proton-conducting unit F₀, forms the H⁺-ATP synthase that reversibly couples transmembrane proton flow to ATP synthesis/hydrolysis in respiring and photosynthetic cells^{9,10}. It has been suggested that the γ -subunit of F₁-ATPase rotates within the $\alpha\beta$ -hexamer¹¹, a conjecture supported by structural⁹, biochemical^{12,13} and spectroscopic¹⁴ studies. We attached a fluorescent actin filament to the γ -subunit as a marker, which enabled us to observe this motion directly. In the presence of ATP, the filament rotated for more than 100 revolutions in an anticlockwise direction when viewed from the 'membrane' side. The rotary torque produced reached more than 40 pN nm⁻¹ under high load.

In the crystal structure of mitochondrial F₁-ATPase⁸, rigid¹⁵ coiled-coil α -helices of the γ -subunit penetrate the central cavity of the $\alpha\beta$, and extend into the stalk region that links F₁-ATPase to the F₀ portion. The amino terminus of the β -subunits is on the side opposite the stalk region of the γ -subunit. To fix the $\alpha\beta\gamma$ subcomplex on a glass plate, the subcomplex derived from a thermophilic bacterium was expressed in *Escherichia coli*, with ten histidines (His tag) linked to the N terminus of each β -subunit. The

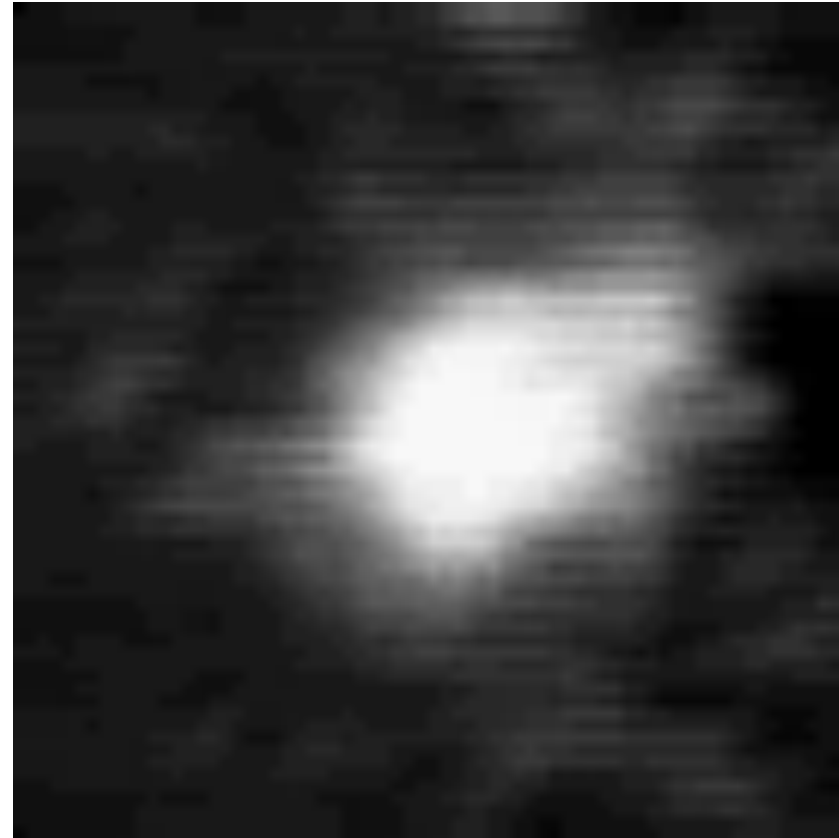


ATP Hydrolysis:
Energy Wasted

H.Noji*, R.Yasuda*, M. Yoshida, and K. Kinoshita, Jr. (1997). *Nature* 386:299-302
Equally contributed

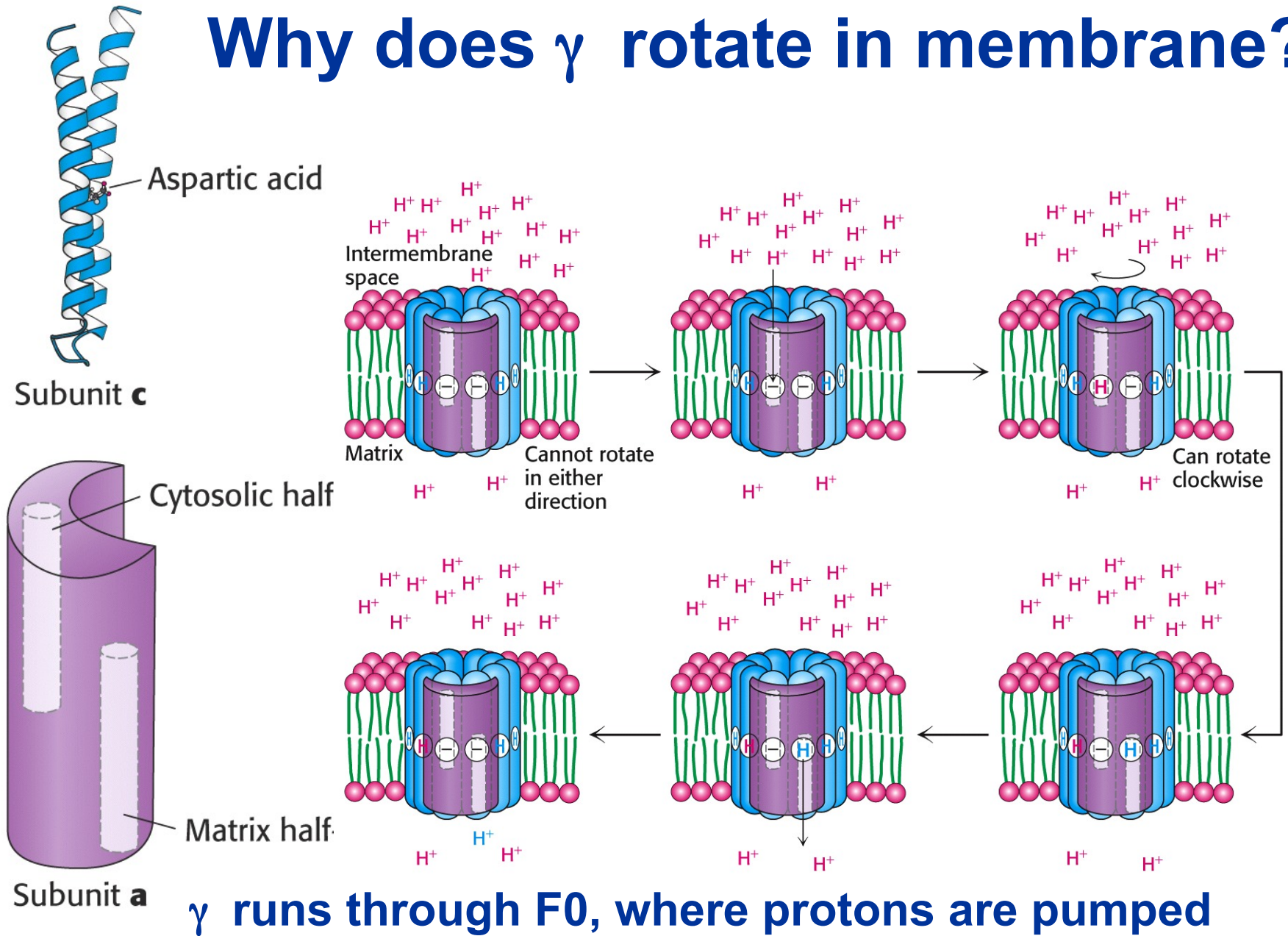
F1 the first single molecule experiments

1. Experimental witness to rotation of γ subunit with single molecule spectroscopy.
2. NO Fo subunit, DRIVING FORCE IS ATP HYDROLYSIS
3. When ATP is hydrolyzed, shaft goes in the opposite direction as before $L \rightarrow O$;
 $T \rightarrow L$: $O \rightarrow T$ releasing ADP product

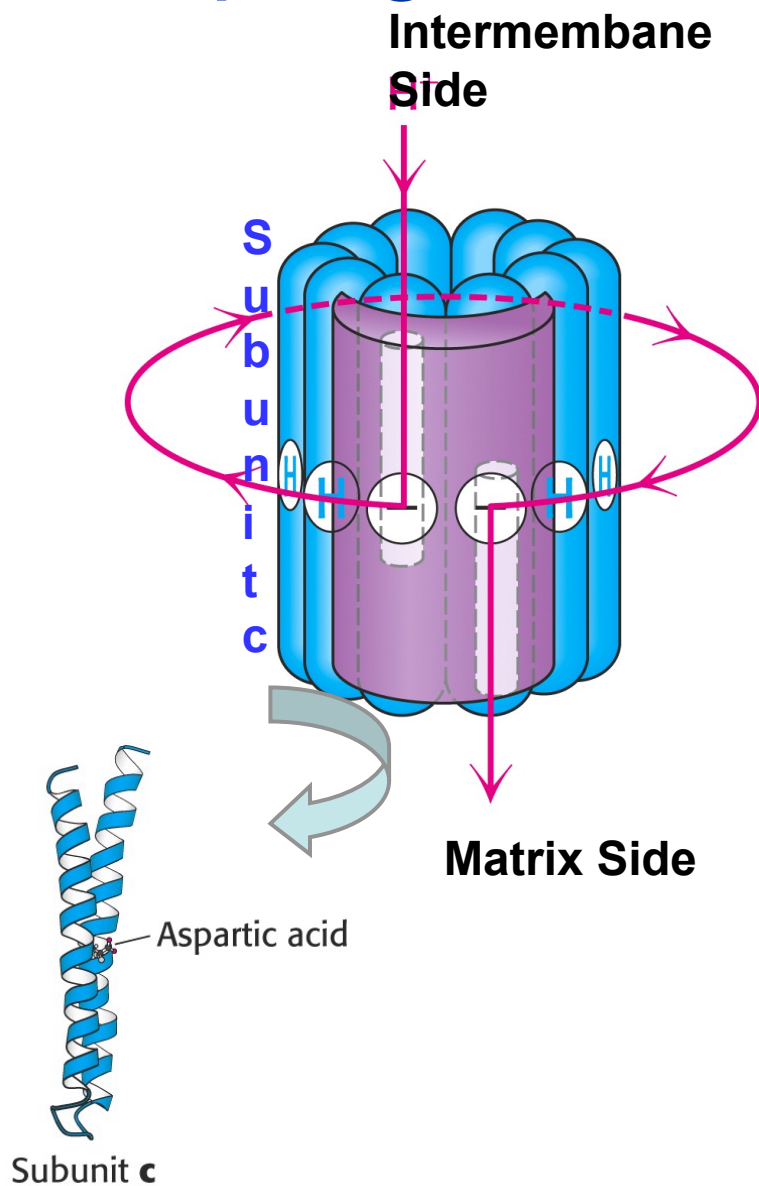


Rotation is in 120° steps
Rotor speed is 100 Hz, 3 ATP/cycle
Rate is 100 ATP/s/subunit,

Why does γ rotate in membrane?

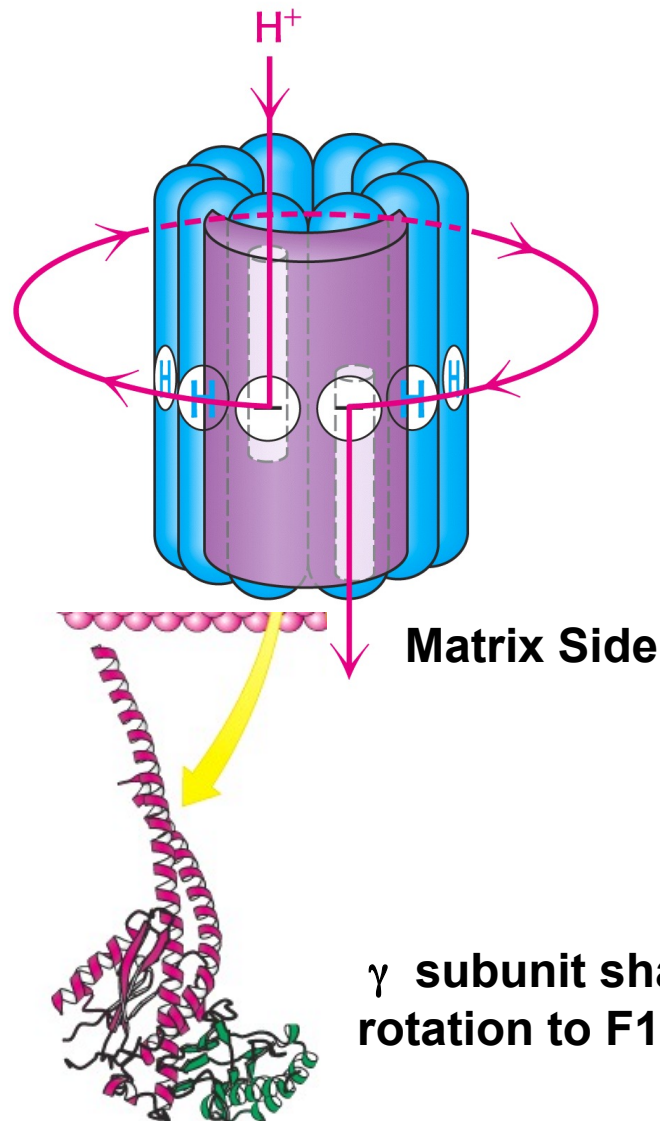


Coupling of Fo to F1



Unlike what seems to be illustrated in this image, it is not the a subunit that rotates, but the c subunits. The c subunits each contain an aspartic acid group centrally located within the helix within the membrane. When subunit a is in association with this subunit c, this asp is connected through a half channel to either the matrix space or the intermembrane space. When the channel connects to the matrix space, the asp deprotonates because of the H^+ gradient. This deprotonation is followed by a conf change which ratchets the newly deprotonated c into alignment with the second half channel, in which it reprotonates from the intermembrane space. Meanwhile, the next c has just deprotonated to matrix, and the assembly ratchets again, moving the newly protonated asp out of alignment with any channel. It will undergo a full rotation around the assembly, driven by the successive protonation/ratchet events, until it comes full circle to the a subunit and the matrix channel, where it once again, deprotonates, ratchets and protonates.

Coupling of Fo to F1



.....Meanwhile, the rotation of the c subunits within Fo transfer this rotation to the γ subunit which acts as a shaft connecting Fo to F1. Rotation of γ about the hexamer of $\alpha_3\beta_3$ converts b subunits successively from O to T to L.

Happy motoring with ATP synthase

Alan E Senior & Joachim Weber

ATP synthase research has reached a milestone as single-molecule techniques are used to examine the direction and stepping of the proton gradient–driven rotation, to determine the effect of forced rotation on ATP synthesis and to synchronously monitor rotation and nucleotide kinetics.

Purposeful generation and utilization of energy define living organisms. The central molecule of biological energetics is ATP, and the enzyme responsible for the majority of ATP synthesis in organisms from bacteria to man is ATP synthase. A rush of excitement ran through this field in 1997 when Noji *et al.*¹ discovered that ATP synthase is not just an enzyme, but also a tiny motor. One part of the enzyme rotates, at ~100 Hz, and rotation is integral to the mechanism—as you read this article, billions of these ‘nanomotors’ are turning at this speed in your tissues and cells. This striking discovery revolutionized experimentation: it was evident that a field that had been previously dominated by membrane biochemistry and enzymology would need to import technologies of physics and biophysics. There

gradient through the enzyme, transducing gradient energy into chemical energy by converting ADP and phosphate (P_i) into ATP.

The ATP synthase from *Escherichia coli*⁵ is composed of two major domains (Fig. 1). The membrane-embedded part (F_o) provides a specific H^+ conduction path, between the c subunit ring and the a subunit. The catalytic sites, three in number, are located at α/β subunit interfaces of the $\alpha_3\beta_3$ hexagon in the membrane-external F_1 . The ‘rotor’ consists of γ and ϵ subunits firmly fastened to the c ring. A helical region of the γ subunit passes through the center of the $\alpha_3\beta_3$ hexagon, providing communication with the catalytic sites. At the periphery, the ‘stator’ ($b_2\delta$) prevents co-rotation of catalytic sites with the rotor due to viscous drag. Visualization of

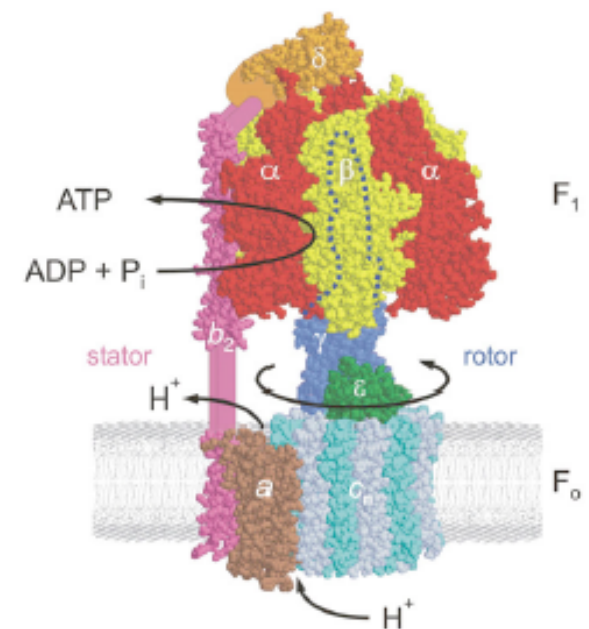


Figure 1 Structure of ATP synthase. The enzyme

C**S**

At low dilution, only one ATP synthase containing liposome will migrate into irradiation volume at a time

ARTICLE

Nature Structural & Molecular Biology **11**, 135 - 141 (2004)
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Proton-powered subunit rotation in single membrane-bound F₀F₁-ATP synthase

Manuel Diez¹, Boris Zimmermann¹, Michael Börsch², Marcelle König³, Enno Schweinberger³, Stefan Steigmiller¹, Rolf Reuter¹, Suren Felekyan³, Volodymyr Kudryavtsev³, Claus A M Seidel³ & Peter Gräber¹

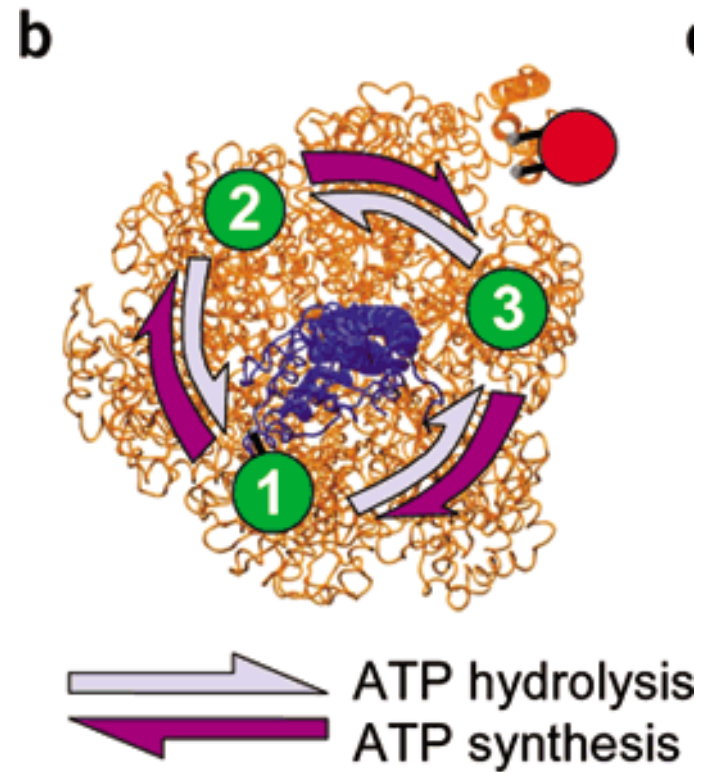
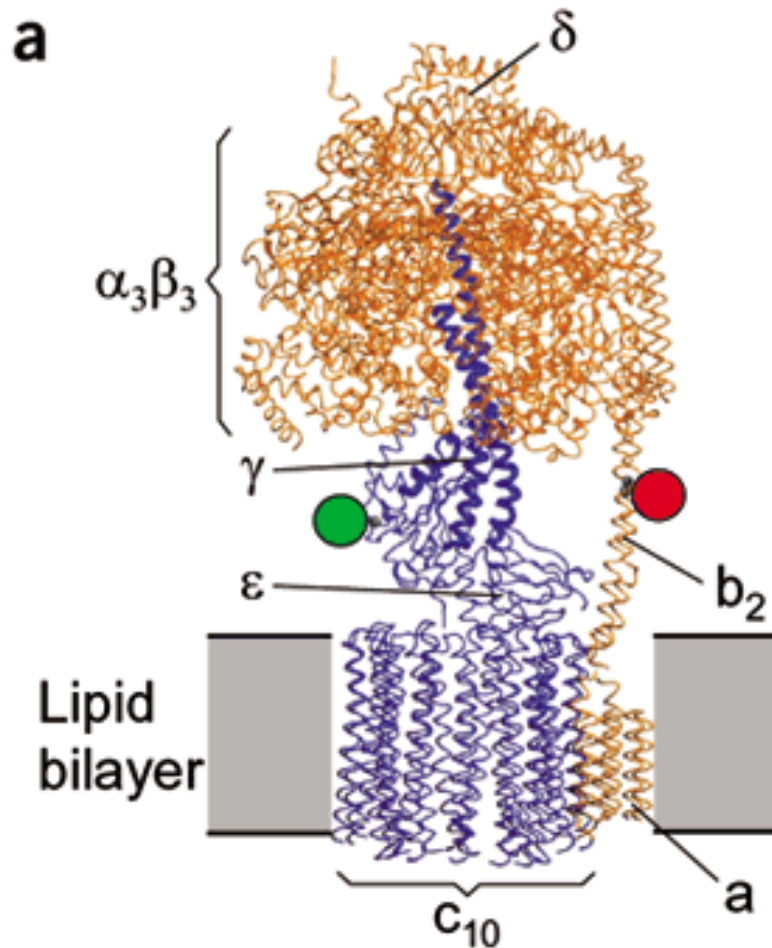
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Synthesis of ATP from ADP and phosphate, catalyzed by F₀F₁-ATP synthases, is the most abundant physiological reaction in almost any cell. F₀F₁-ATP synthases are membrane-bound enzymes that use the energy derived from an electrochemical proton gradient for ATP formation. We incorporated double-labeled F₀F₁-ATP synthases from *Escherichia coli* into liposomes and measured single-molecule fluorescence resonance energy transfer (FRET) during ATP synthesis and hydrolysis. The γ subunit rotates stepwise during proton transport-powered ATP synthesis, showing three distinct distances to the b subunits in repeating sequences. The average durations of these steps correspond to catalytic turnover times upon ATP synthesis as well as ATP hydrolysis. The direction of rotation during ATP synthesis is opposite to that of ATP hydrolysis.



Green sphere is fluorescence donor on γ (which rotates) and red sphere is fluorescence acceptor on stator, which doesn't move. In a FRET experiment, whenever the donor (green) comes close to the acceptor (red) energy will be transferred to acceptor.

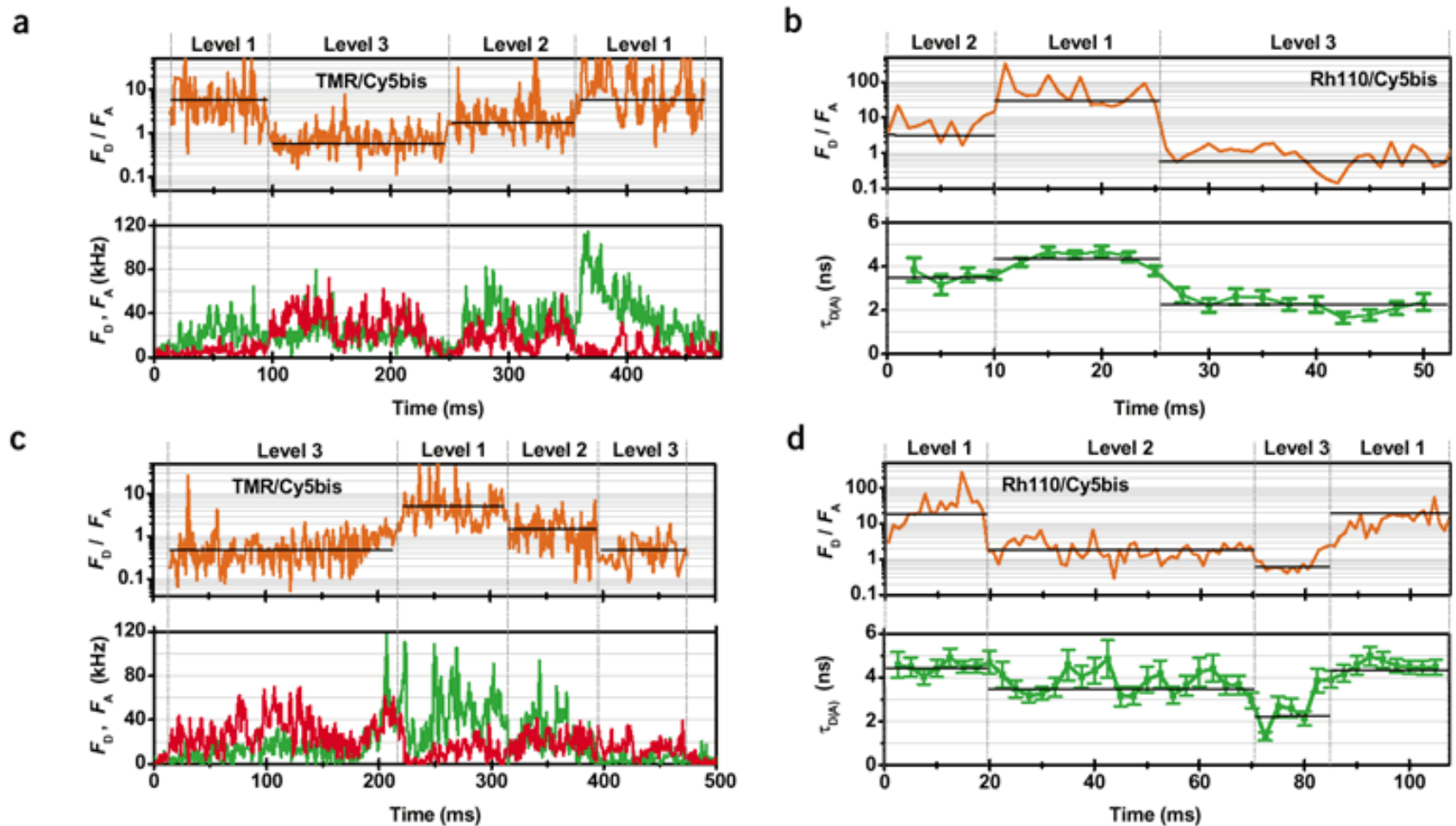


Figure 2. Photon bursts from single F₀F₁-ATP synthases in liposomes.

(a,b) Photon bursts during ATP hydrolysis. (c,d) Photon bursts during ATP synthesis. FRET donor is TMR in a and c, Rh110 in b and d; FRET acceptor is Cy5bis in all traces. Fluorescence intensity of the donor, F_D , and acceptor, F_A , are green and red, respectively, in a and c (time window 1 ms). Corrected intensity ratios F_D / F_A are orange in all panels (time window 1 ms), fluorescence lifetimes of the donor Rh110 are green in b and d (total time window 5 ms, shifted by 2.5 ms per data point). Three distinct FRET levels are attributed (1, 2 or 3) at top of panels.

Mechanically driven ATP synthesis by F_1 -ATPase

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⁵Institute of Industrial Science, University of Tokyo, Tokyo 153-8505, Japan
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⁷ERATO "ATP System", 5800-3 Nagatsuta, Yokohama 226-0026, Japan

ATP, the main biological energy currency, is synthesized from ADP and inorganic phosphate by ATP synthase in an energy-requiring reaction¹⁻³. The F_1 portion of ATP synthase, also known as F_1 -ATPase, functions as a rotary molecular motor: *in vitro* its γ -subunit rotates⁴ against the surrounding $\alpha_3\beta_3$ subunits⁵, hydrolysing ATP in three separate catalytic sites on the β -subunits. It is widely believed that reverse rotation of the γ -subunit, driven by proton flow through the associated F_0 portion of ATP synthase, leads to ATP synthesis in biological systems^{1-3,6,7}. Here we present direct evidence for the chemical synthesis of ATP driven by mechanical energy. We attached a magnetic bead to the γ -subunit of isolated F_1 on a glass surface, and rotated the bead using electrical magnets. Rotation in the appropriate direction resulted in the appearance of ATP in the medium as detected by the luciferase-luciferin reaction. This shows that a vectorial force (torque) working at one particular point on a protein machine can influence a chemical reaction occurring in physically remote catalytic sites, driving the reaction far from equilibrium.

When isolated F_1 hydrolyses ATP, its central γ -subunit rotates

